

Monatomic Ions

The charge of a monatomic ion can often be determined by the element's position on the periodic table:

Group 1A: 1+ Ex. Na^+

Group 2A: 2+ Ex. Ca^{2+}

Group 3A: 3+ Ex. Al^{3+}

Group 5A: 3- Ex. N^{3-}

Group 6A: 2- Ex. O^{2-}

Group 7A: 1- Ex. F^-

Monatomic Ion 1 Answers

Denise Guinn, Rebecca Brewer



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