

METALLIC BONDS



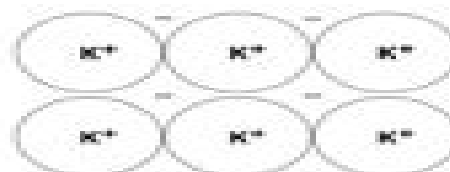
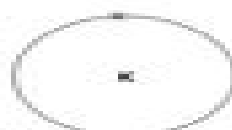
Draw diagrams to represent bonding in the each of the metals 1-5. Use six particles in each diagram.

1. Sodium, Na



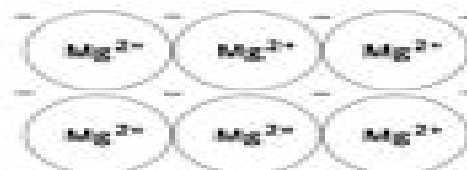
6 ions and 6 electrons

2. Potassium, K



6 ions and 6 electrons

3. Magnesium, Mg



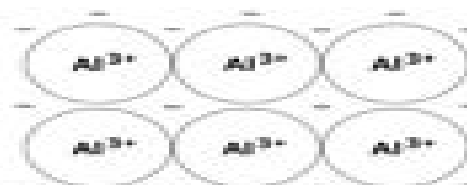
6 ions and 12 electrons

4. Calcium, Ca



6 ions and 12 electrons

5. Aluminium, Al



6 ions and 18 electrons

Metallic Bonding Ws Answers

LL Leslie



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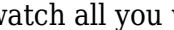
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