

# HYDROCARBONS



## Hydrocarbon Compounds

Hydrocarbon compounds are compounds that consist only of carbon and hydrogen atoms. Hydrocarbons can be divided into 2, namely:

### a. Saturated Hydrocarbons: single bonds

including saturated hydrocarbons :

- Alkane : open chain ( $C_nH_{2n+2}$ )
- Cycloalkane : closed chain ( $C_nH_{2n}$ )

### b. Unsaturated Hydrocarbons: double bonds

- Alkene : double bond ( $C_nH_{2n}$ )
- Alkyne : triple bond ( $C_nH_{2n-2}$ )
- Alkadiene : two double bonds ( $C_nH_{2n-2}$ )



## Nomenclature of Alkanes

The nomenclature of alkanes is the basis for the nomenclature of other carbon compounds. Therefore the nomenclature of alkanes must be thoroughly understood.

Here are the names of alkanes from  $C_1$ - $C_{10}$  :

Formula	Name	Formula	Name
$CH_4$	methane	$C_6H_{14}$	hexane
$C_2H_6$	ethane	$C_7H_{16}$	heptane
$C_3H_8$	propane	$C_8H_{18}$	octane
$C_4H_{10}$	butane	$C_9H_{20}$	nonane
$C_5H_{12}$	pentane	$C_{10}H_{22}$	decane

Alkene names :

$C_2H_4$  = ethylene

$C_3H_6$  = propene

# Hydrocarbons Note Taking Guide

**M Carnoy**



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## **Table of Contents Hydrocarbons Note Taking Guide**

1. Understanding the eBook Hydrocarbons Note Taking Guide
  - The Rise of Digital Reading Hydrocarbons Note Taking Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Hydrocarbons Note Taking Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hydrocarbons Note Taking Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Hydrocarbons Note Taking Guide
  - Personalized Recommendations
  - Hydrocarbons Note Taking Guide User Reviews and Ratings

- Hydrocarbons Note Taking Guide and Bestseller Lists
- 5. Accessing Hydrocarbons Note Taking Guide Free and Paid eBooks
  - Hydrocarbons Note Taking Guide Public Domain eBooks
  - Hydrocarbons Note Taking Guide eBook Subscription Services
  - Hydrocarbons Note Taking Guide Budget-Friendly Options
- 6. Navigating Hydrocarbons Note Taking Guide eBook Formats
  - ePub, PDF, MOBI, and More
  - Hydrocarbons Note Taking Guide Compatibility with Devices
  - Hydrocarbons Note Taking Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Hydrocarbons Note Taking Guide
  - Highlighting and Note-Taking Hydrocarbons Note Taking Guide
  - Interactive Elements Hydrocarbons Note Taking Guide
- 8. Staying Engaged with Hydrocarbons Note Taking Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hydrocarbons Note Taking Guide
- 9. Balancing eBooks and Physical Books Hydrocarbons Note Taking Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hydrocarbons Note Taking Guide
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Hydrocarbons Note Taking Guide
  - Setting Reading Goals Hydrocarbons Note Taking Guide
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hydrocarbons Note Taking Guide
  - Fact-Checking eBook Content of Hydrocarbons Note Taking Guide
  - Distinguishing Credible Sources

### 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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