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# CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

## Matter—Properties and Changes

### Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

|                     |                    |                       |
|---------------------|--------------------|-----------------------|
| chemical<br>density | mass<br>properties | physical<br>substance |
|---------------------|--------------------|-----------------------|

Matter is anything with (1) mass and volume. A

(2) substance is a form of matter with a uniform and unchanging composition.

Substances have specific, unchanging (3) properties that can be observed.

Substances have both physical and chemical properties. (4) Physical

properties can be observed without changing a substance's chemical composition. Color,

hardness, and (5) density are examples. Other properties cannot be

observed without changing the composition of a substance. These are called

(6) chemical properties. An example is the tendency of iron to form

rust when exposed to air.

Label each property as either physical or chemical.

- physical 7. Chemical formula  $H_2O$
- chemical 8. Irons react vigorously when exposed to moist air
- chemical 9. Remains unchanged when in the presence of nitrogen
- physical 10. Colorless
- physical 11. Solid at normal temperatures and pressures
- chemical 12. Ability to combine with another substance
- physical 13. Melting point
- physical 14. Liquid at normal temperatures and pressures
- physical 15. Boiling point is  $100^{\circ}C$
- physical 16. Conducts electricity
- physical 17. Density is  $\frac{1}{2} \frac{g}{cm^3}$

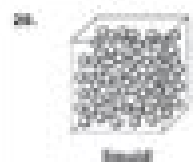
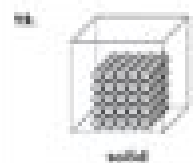
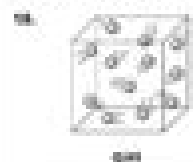
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# CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

### Section 3.1 continued

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



For each statement below, write true or false.

- true 21. All matter that we encounter in everyday life exists in one of these physical forms.
- true 22. A solid has definite shape and volume.
- false 23. A liquid has a definite shape and takes on the volume of its container.
- true 24. A gas has both the shape and the volume of its container.
- false 25. The particles in a gas cannot be compressed into a smaller volume.
- false 26. Liquids tend to contract when heated.
- false 27. The particles in a solid are spaced far apart.
- false 28. The words gas and vapor can be used interchangeably.

# Matter And Change Chapter 8 2 Study Guide Key

**N Colangelo**



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## **Table of Contents Matter And Change Chapter 8 2 Study Guide Key**

1. Understanding the eBook Matter And Change Chapter 8 2 Study Guide Key
  - The Rise of Digital Reading Matter And Change Chapter 8 2 Study Guide Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Matter And Change Chapter 8 2 Study Guide Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Matter And Change Chapter 8 2 Study Guide Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Matter And Change Chapter 8 2 Study Guide Key
  - Personalized Recommendations
  - Matter And Change Chapter 8 2 Study Guide Key User Reviews and Ratings
  - Matter And Change Chapter 8 2 Study Guide Key and Bestseller Lists
5. Accessing Matter And Change Chapter 8 2 Study Guide Key Free and Paid eBooks

- Matter And Change Chapter 8 2 Study Guide Key Public Domain eBooks
  - Matter And Change Chapter 8 2 Study Guide Key eBook Subscription Services
  - Matter And Change Chapter 8 2 Study Guide Key Budget-Friendly Options
6. Navigating Matter And Change Chapter 8 2 Study Guide Key eBook Formats
- ePub, PDF, MOBI, and More
  - Matter And Change Chapter 8 2 Study Guide Key Compatibility with Devices
  - Matter And Change Chapter 8 2 Study Guide Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Matter And Change Chapter 8 2 Study Guide Key
  - Highlighting and Note-Taking Matter And Change Chapter 8 2 Study Guide Key
  - Interactive Elements Matter And Change Chapter 8 2 Study Guide Key
8. Staying Engaged with Matter And Change Chapter 8 2 Study Guide Key
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Matter And Change Chapter 8 2 Study Guide Key
9. Balancing eBooks and Physical Books Matter And Change Chapter 8 2 Study Guide Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Matter And Change Chapter 8 2 Study Guide Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Matter And Change Chapter 8 2 Study Guide Key
- Setting Reading Goals Matter And Change Chapter 8 2 Study Guide Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Matter And Change Chapter 8 2 Study Guide Key
- Fact-Checking eBook Content of Matter And Change Chapter 8 2 Study Guide Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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