

Key

Ideal Gas Law Worksheet $PV = nRT$

Use the ideal gas law, " $PerV=nRT$ ", and the universal gas constant $R = 0.0821 \frac{L \cdot atm}{K \cdot mol}$ to solve the following problems:

If pressure is needed in kPa then convert by multiplying by $101.3 kPa / 1 atm$ to get
 $R = 8.31 kPa \cdot L / (K \cdot mole)$

- 1) If I have 4 moles of a gas at a pressure of 5.6 atm and a volume of 12 liters, what is the temperature?

$$PV = nRT$$

$$T = \frac{PV}{nR} = \frac{(5.6 atm)(12 L)}{4 mol \cdot 0.0821 \frac{L \cdot atm}{K \cdot mol}}$$

$$T = 204.63 K$$

- 2) If I have an unknown quantity of gas at a pressure of 1.2 atm, a volume of 31 liters, and a temperature of $87^{\circ}C$, how many moles of gas do I have?

$$PV = nRT$$

$$n = \frac{PV}{RT} = \frac{1.2 atm \cdot 31 L}{0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 360 K}$$

$$n = 1.2586 mol$$

- 3) If I contain 3 moles of gas in a container with a volume of 60 liters and at a temperature of 400 K, what is the pressure inside the container?

$$PV = nRT$$

$$P = \frac{nRT}{V} = \frac{3 mol \cdot 0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 400 K}{60 L}$$

$$P = 1.642 atm$$

$$or \quad P = 166.29 kPa$$

- 4) If I have 7.7 moles of gas at a pressure of 0.09 atm and at a temperature of $56^{\circ}C$, what is the volume of the container that the gas is in?

$$PV = nRT$$

$$V = \frac{nRT}{P} = \frac{7.7 mol \cdot 0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 327 K}{0.09 atm}$$

$$V = 2310.93 L$$

- 5) If I have 17 moles of gas at a temperature of $67^{\circ}C$, and a volume of 88.89 liters, what is the pressure of the gas?

$$PV = nRT$$

$$P = \frac{nRT}{V} = \frac{17 mol \cdot 0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 390 K}{88.89 L}$$

$$P = 5.34 atm$$

$$or \quad P = 540.61 kPa$$

- 6) If I have an unknown quantity of gas at a pressure of 0.5 atm, a volume of 25 liters, and a temperature of 300 K, how many moles of gas do I have?

$$PV = nRT$$

$$n = \frac{PV}{RT} = \frac{0.5 atm \cdot 25 L}{0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 300 K}$$

$$n = 0.5075 mol$$

Ideal Gas Law Problems Answer Key

**Richard W. Hartel, D.B. Hyslop, Terry
Howell, Jr.**



Ideal Gas Law Problems Answer Key:

Math Concepts for Food Engineering Richard W. Hartel, D.B. Hyslop, Terry Howell, Jr., 1997-05-19 Food industry managers engineers and scientists require a personal repertory of mathematical tools to find solutions to a variety of work problems These problems relate to processing research analysis and other operations that require mathematical solutions This new handbook provides a practical review of required basic mathematical methods and relates them to specific areas of food industry problem solving Prepared for students in undergraduate food science programs the handbook also serves the needs of food industry technical personnel who would be helped by a review of basic math for food engineering and non technical managers who seek an introduction to food engineering math and problem solving Each chapter on a math or problem solving skill begins with fundamental concepts presents step by step procedures in the skill and provides numerous practical illustrative examples Practice problems with answers in the appendix provide for self testing and reinforcement of learning All of the math skills presented relate to common food industry problems From the text In food engineering and any other mathematically oriented discipline one must become adept at problem solving using equations and mathematical principles Many of the problems that food scientists and engineers face require the ability to apply physical and mathematical principles to solve equations and calculate necessary information Thus problem solving abilities are extremely important and we should be familiar with the general steps in problem solving Investigating Chemistry Matthew Johll, 2008-12-22 In its new second edition Investigating Chemistry A Forensic Science Perspective remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways that are effective and accessible for students This edition features refined explanations of the chemical concepts which are the core of the book as well as a more thoroughly integrated forensic theme updated features and an expanded media supplements package Vol 14: Thermodynamics: Adaptive Problems Book in Physics (with Detailed Solutions) for College & High School SATYAM SIR, 2021-08-01 Learn Thermodynamics which is divided into various sub topics Each topic has plenty of problems in an adaptive difficulty wise From basic to advanced level with gradual increment in the level of difficulty The set of problems on any topic almost covers all varieties of physics problems related to the chapter Thermodynamics If you are preparing for IIT JEE Mains and Advanced or NEET or CBSE Exams this Physics eBook will really help you to master this chapter completely in all aspects It is a Collection of Adaptive Physics Problems in Thermodynamics for SAT Physics AP Physics 11 Grade Physics IIT JEE Mains and Advanced NEET Olympiad Level Book Series Volume 14 This Physics eBook will cover following Topics for Thermodynamics 1 Ideal Gas Equation 2 Thermodynamic Processes 3 1st Law of Thermodynamics 4 Graphs 5 Polytrophic Process 6 Cyclic

Process 7 2nd Law of Thermodynamics Heat Engine 8 2nd Law of Thermodynamics Heat Pump 9 Chapter Test The intention is to create this book to present physics as a most systematic approach to develop a good numerical solving skill About Author Satyam Sir has graduated from IIT Kharagpur in Civil Engineering and has been teaching Physics for JEE Mains and Advanced for more than 8 years He has mentored over ten thousand students and continues mentoring in regular classroom coaching The students from his class have made into IIT institutions including ranks in top 100 The main goal of this book is to enhance problem solving ability in students Sir is having hope that you would enjoy this journey of learning physics In case of query visit www.physicsfactor.com or WhatsApp to our customer care number 91 7618717227 **Chemical**

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Vol 05: Heat & Thermodynamics : Adaptive Problems Book in Physics for College & High School

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