CASCALLER AND PROPERTY.

(There is no Prelaboratory Assignment for this exercise. Make entries in your lab notebook for anything that is an observation; for your report, turn in this page and the following page.)

## Making molecular models

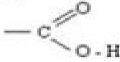
Make models of the following functional groups (additional red O balls will be needed):

make one C-C group

make four -H

make two C=C groups

make four



groups

Make models of the following dicarboxylic acids and observe:

- A. Rotation around chemical bonds:
  - Make succinic acid, HOOC-CH2-CH2-COOH.

Does the molecule rotate around the central C-C bond?

2.Replace the central CH2-CH2 group with a HC=CH group (keep one COOH group on each end).

Does the molecule rotate around the central C=C bond?

- 3. What is the molecular formula (C<sub>i</sub>H<sub>j</sub>O<sub>k</sub>)of your new molecule?
- B. Comparison of maleic and fumaric acids.

Maleic acid and fumaric acid have the same molecular formula (C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>) and the same condensed structural formula (HOOC-CH=CH-COOH). Your model is either maleic or fumaric acid.

- Compare your model with others in the class. What differences do you see?
- 2. Make a second model of the other form.

# **Geometric Isomers Lab Answers**

**RJ Shavelson** 

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- 1. Understanding the eBook Geometric Isomers Lab Answers
  - The Rise of Digital Reading Geometric Isomers Lab Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Geometric Isomers Lab Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Geometric Isomers Lab Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Geometric Isomers Lab Answers
  - Personalized Recommendations
  - Geometric Isomers Lab Answers User Reviews and Ratings
  - Geometric Isomers Lab Answers and Bestseller Lists

- 5. Accessing Geometric Isomers Lab Answers Free and Paid eBooks
  - Geometric Isomers Lab Answers Public Domain eBooks
  - Geometric Isomers Lab Answers eBook Subscription Services
  - Geometric Isomers Lab Answers Budget-Friendly Options
- 6. Navigating Geometric Isomers Lab Answers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Geometric Isomers Lab Answers Compatibility with Devices
  - Geometric Isomers Lab Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Geometric Isomers Lab Answers
  - Highlighting and Note-Taking Geometric Isomers Lab Answers
  - Interactive Elements Geometric Isomers Lab Answers
- 8. Staying Engaged with Geometric Isomers Lab Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Geometric Isomers Lab Answers
- 9. Balancing eBooks and Physical Books Geometric Isomers Lab Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Geometric Isomers Lab Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Geometric Isomers Lab Answers
  - Setting Reading Goals Geometric Isomers Lab Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Geometric Isomers Lab Answers
  - Fact-Checking eBook Content of Geometric Isomers Lab Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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