

**Heat & Gases**  
**Chapter 2 - Heat and Internal Energy**  
**Exercise**

**Multiple Choice (MC)**

1. Which of the following statements is/are correct?  
(1) Hot objects store more heat than cold objects  
(2) Average molecular kinetic energy is minimum at absolute zero  
(3) Molecules of a solid can move freely  
  
A. (2) only                      B. (1) and (2) only                      C. (1) and (3) only  
D. (1), (2) and (3)
  
2. Which of the following statements is/are correct?  
(1) 1 kg of water at  $50^{\circ}\text{C}$  has same average molecular kinetic energy as 2 kg of water at  $25^{\circ}\text{C}$   
(2) 1 kg of ice has smaller average molecular potential energy than 0.1 kg of steam  
(3) When water vaporizes, it releases energy to the surroundings  
  
A. (2) only                      B. (3) only                      C. (1) and (2) only  
D. (1), (2) and (3)
  
3. Which of the following statements is/are NOT correct?  
(1) Hotter objects have greater internal energy than colder objects  
(2) Transfer of heat only occurs when two objects are in contact  
(3) Energy always flows from an object with greater average molecular kinetic energy to an object with smaller average molecular kinetic energy  
  
A. (1) only                      B. (3) only                      C. (1) and (2) only  
D. (1) and (3) only
  
4. A heater releases 280 kJ of heat in 7 minutes. What is its power?  
  
A. 4 W                      B. 40 W                      C. 66.7 W                      D. 667 W

# Multiple Choice Questions On Heat Energy

**CH Cherryholmes**



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