

STUDY GUIDE: Understanding Properties of Matter

1. Use examples to describe the difference between physical and chemical properties of matter. (See handout)
2. List 9 physical properties of matter with an example of each. **Density, Ductility, Electrical Conductivity, Magnetism, Malleability, Solubility, Specific Heat, State, Thermal Conductivity** (Examples will vary but should be similar to the examples found in the Properties of Matter Booklet.)
3. List 2 chemical properties of matter with an example of each. **Flammability, Reactivity** (Examples will vary but should be similar to the examples found in the Properties of Matter Booklet.)
4. Which property of matter is observed when water changes from a solid to a liquid? **State**
5. Adding food coloring to the solution is an example of **solubility**.
6. If bubbles start to rise from a solution in the scientist's test tube shown, it would be an indication of **reactivity**.
7. What "point" is observed when water changes from a solid to a liquid? **Melting point**
8. Which property(ies) would you use to describe most metals? **Ductility or Malleability or Electrical Conductivity**
9. Which of the properties of metals means that it can be drawn into wires? **Ductility**
10. Which property allows metals to be bent rather than broken? **Malleability**
11. Besides being valuable, gold is special because it can be hammered into a very thin foil. This property is called **malleability**.

Use the table below to interpret questions 12 – 15.

SUBSTANCE	DENSITY (g/cm ³)	SUBSTANCE	DENSITY (g/cm ³)
Aluminum (solid)	2.7	Ice (solid)	0.93
Pyrite (solid)	5.02	Water (liquid)	1.00
Mercury (liquid)	13.55	Zinc (solid)	7.13
Silver (solid)	10.50	Wood (oak)	0.85

12. What substance has a density more than 13 times greater than water? **Mercury**
13. What is the density of ice? **.93g/cm³**
14. Which metal is a liquid at room temperature? **Mercury**
15. What two variables determine the density of a substance? **Mass/volume**
16. The electromagnets found inside of doorbells are an example of which properties of matter? **Magnetism; Electrical conductivity; Ductility (coiled wire around metal)**
17. Chemical weathering, such as acid rain causing a statue to fizz and deteriorate is an example of which property of matter? **Reactivity**
18. Which property of matter is exemplified by pressing metals into coins? **Malleability**
19. Which property of matter is observed when water changes from a liquid to a solid? **State**
20. What "point" is observed when water changes from a liquid to a solid? **Freezing point**
21. Which property of matter is observed when water changes from a liquid to a gas? **State**
22. What "point" is observed when water changes from a liquid to a gas? **Boiling point**

Matter Of States Viewing Guide Key

LL Leslie



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Table of Contents Matter Of States Viewing Guide Key

1. Understanding the eBook Matter Of States Viewing Guide Key
 - The Rise of Digital Reading Matter Of States Viewing Guide Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Matter Of States Viewing Guide Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Matter Of States Viewing Guide Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Matter Of States Viewing Guide Key
 - Personalized Recommendations
 - Matter Of States Viewing Guide Key User Reviews and Ratings

- Matter Of States Viewing Guide Key and Bestseller Lists
- 5. Accessing Matter Of States Viewing Guide Key Free and Paid eBooks
 - Matter Of States Viewing Guide Key Public Domain eBooks
 - Matter Of States Viewing Guide Key eBook Subscription Services
 - Matter Of States Viewing Guide Key Budget-Friendly Options
- 6. Navigating Matter Of States Viewing Guide Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Matter Of States Viewing Guide Key Compatibility with Devices
 - Matter Of States Viewing Guide Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Matter Of States Viewing Guide Key
 - Highlighting and Note-Taking Matter Of States Viewing Guide Key
 - Interactive Elements Matter Of States Viewing Guide Key
- 8. Staying Engaged with Matter Of States Viewing Guide Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Matter Of States Viewing Guide Key
- 9. Balancing eBooks and Physical Books Matter Of States Viewing Guide Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Matter Of States Viewing Guide Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Matter Of States Viewing Guide Key
 - Setting Reading Goals Matter Of States Viewing Guide Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Matter Of States Viewing Guide Key
 - Fact-Checking eBook Content of Matter Of States Viewing Guide Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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