

ICSE Class 10 Chemistry Question Paper Solution 2017

CHEMISTRY (SCIENCE PAPER-2)

SECTION I (40 Marks)

Attempt all questions from this Section

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Ju	esti	on 1	
(a)	Fill	in the blanks from the choices given in brackets:	[5]
	(i)	The energy required to remove an electron from a neutral isolated gaseous atom and convert it into a positively charged gaseous ion is called (electron affinity, ionisation potential, electronegativity)	
	(iii)	The compound that does not have a lone pair of electrons is (water, ammonia, carbon tetra chloride)	
	(iii)	When a metallic oxide is dissolved in water, the solution formed has a high concentration ofions. (H^+, H_2O^+, OH^-)	
	(iv)	Potassium sulphite on reacting with hydrochloric acid releasesgas. (Cl ₂ , SO ₂ , H ₂ S)	
	(v)	The compound formed when ethene reacts with Hydrogen is (CH_4 , C_2H_6 , C_3H_8)	
(b)	Cho	ose the correct answer from the options given below:	[5]
	(i)	A chloride which forms a precipitate that is soluble in excess of ammonium hydroxide, is:	
		1.Calcium chloride	
		2.Ferrous chloride	
		3. Ferric chloride	
		4. Copper chloride	
	(ii)	If the molecular formula of an organic compound is C10H18 it is:	

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