

Name _____ Date _____

Chapter 8 Study Guide Covalent Bonding

- What is a covalent bond?
- What are the characteristics of molecular compounds (covalently bonded compounds)?

Characteristics of Ionic and Covalent Compounds		
Characteristics	Ionic compound	Covalent compound
Representative element	Formula unit	
Bond formation	Transfer of one or more electrons between atoms	
Type of elements	Metallic and nonmetallic	
Physical state	Solid	
Melting point	High (usually above 300 deg. Celcius)	
Solubility in water	Usually high	
Electrical conductivity of aqueous solution	Good conductor	

- Know how to write the Lewis electron dot structure for a molecular compound given its molecular formula.
- The trick part is figuring out how many bonds to draw. It's easy if you follow these steps.

1. Count the number of _____
2. Subtract that number from _____
3. Divide your answer by _____
4. That number equals _____
5. Draw the structure with the correct number of bonds and don't forget to make sure every octet is full.

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Ian Pickup



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