

CANDIDATE NAME:

CLASS:



MINISTRY OF EDUCATION
NGASO EDUCATION CENTRE



MID TERM TWO EXAMINATION GRADE 10 - 2025

CHEMISTRY

Time: 1:30hrs

SECTION A

1. Name a branch of chemistry that deals with the study of the physical principles that underline the structure of matter and chemical transformations.
A. Analytical chemistry. B. Biochemistry. C. Organic chemistry. D. Physical chemistry
2. Which of the following is the basic unit of matter?
A. Compounds. B. Atoms. C. Mixtures. D. Elements
3. What do we call a reaction where heat energy is given out from the system to the surrounding?
A. Condensation reaction. B. Decomposition reaction
C. Exothermic reaction D. Endothermic reaction
4. Among the following factors, which one can affect the rate of diffusion?
A. Pressure B. High pressure
C. Low temperature. D. Rate of reaction
5. Which state(s) of matter exist(s) at the freezing point of a substance?
A. Solid only B. Solid and liquid C. Liquid only D. liquid and gas
6. Air is used to inflate tyres because
A. it is readily compressed. B. its molecules move randomly.
C. it serves as a coolant D. it get into tyres faster than other substances
7. An endothermic reaction is one that
A. Evolves heat B. Produces light C. Absorbs energy D. produces sound
8. A teacher asked Mutinta to explain what happens to the particles in a stone when it is heated. The correct explanation given by Mutinta was particles in a stone ...
A. Will not move C. vibrate more in their fixed positions
B. Move randomly D. vibrate and begin to move randomly
9. Which of the following can be used to measure a volume of 15.6 cm accurately?
A. Burette B. Pipette C. Measuring cylinder D. Volumetric flask
10. Diffusion occurs more quickly in gas than in liquid because
A. Molecules in a gas have more frequent collisions than those in liquid.
B. Gas molecules are larger.
C. Gas molecules move randomly
D. On average molecules in a gas are further apart than those in a liquid

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