Inorganic Chemistry 2 Exam Review With 100% Correct Answers 2024

| 1) Which of the following statements about gases is false? |
|---|
| A) All gases are colorless and odorless at room temperature. |
| B) Distances between molecules of gas are very large compared to bond distances within molecules. |
| C) Gases expand spontaneously to fill the container they are placed in. |
| D) Gases are highly compressible. |
| E) Non-reacting gas mixtures are homogeneous answer√√A) All gases are colorless and odorless at room temperature. |
| The dissolution of water in octane (C8H18) is principally prevented by |
| A) dipole-dipole attraction between octane molecules |
| B) repulsion between like -charged water and octane molecules |
| C) hydrogen bonding between water molecules |
| D) London dispersion forces between octane molecules |

Inorganic Chemistry Practice Exam Answers

Léo Pomar

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