# HSC Maharashtra Board question paper: March 2013

#### Note:

- All question are compulsory
- ii. Answer to the two sections are to be written in the same answer book.
- iii. Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Encery new question must be started on a new page.
- vi. Use of logarithmic table is allowed

# CHEMISTRY: SECTION - I

#### Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7

| i. Ir  | n body centred cubic structure the space occupied is about                   |   |                   |     |                                       |
|--------|--|---|-------------------|-----|---------------------------------------|
| (2     | (A)  | 68 %  |                   | (B) | 53 %                                  |
| (0     | C).  | 38 %  |                   | (D) | 32 %                                  |
| ii. Fe | For a gaseous reaction the unit of rate of reaction is                       |   |                   |     |                                       |
| (2     | A)   | L atm s <sup>-1</sup>                                 |                   | (B) | arm mol <sup>-1</sup> s <sup>-1</sup> |
| (6     | C)   | atm s 1   |                   | (D) | mol s                                 |
| iii. W | Which of the following compounds contains S = O as well as S = S bonds?      |   |                   |     |                                       |
| C2     | (A)  | Sulphuric acid  |                   | (B) | Thiosulphuric acid                    |
| (6     | C)   | Sulphurous acid                                       | 00                | (D) | Thiosulphurous acid                   |
| iv. W  | Which of the following solutions shows maximum depression in freezing point? |   |                   |     |                                       |
| (4     | (A)  | 0.5 M Li <sub>2</sub> SO <sub>4</sub>                 |                   | (B) | 1 M NaCl                              |
| 0      | C)   | 0.5 M Al <sub>2</sub> (SO <sub>4</sub> ) <sub>5</sub> | ) "               | (D) | 0.5 M BaCl <sub>2</sub>               |
| v. F   | For a chemical reaction $\Delta S = -0.035$ kJ/K and $\Delta H = -20$ kJ.    |   |                   |     |                                       |
| (4     | A)   | 5.14 K  |                   | (B) | 57.14 K                               |
| 6      | 2)   | 571.4 K   |                   | (D) | 5714.0 K                              |
| vi. T  | The standard e.m.f of the following cell is 0.463 V                          |   |                   |     |                                       |
| C      | Su   | Cum   Agan   Ag. I                                    | If Eas - 0.800 V, |     |                                       |

vii. Fe<sub>2</sub>O<sub>3</sub> is reduced to spongy iron near the top of blast furance by

(A) H<sub>2</sub>

(A) 1.137 V

(C) 0.463 V

(B) CaO

0.337 V

-0.463 V

(B)

(ID)

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What is the standard potential of Cu electrode?

# **Hsc Chemistry Exam 2013 Objective Answer**

**Michael Rajnik** 

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