

GENERAL INORGANIC CHEMISTRY

Study of Matter

A. Classification of Matter

1. Pure Substance

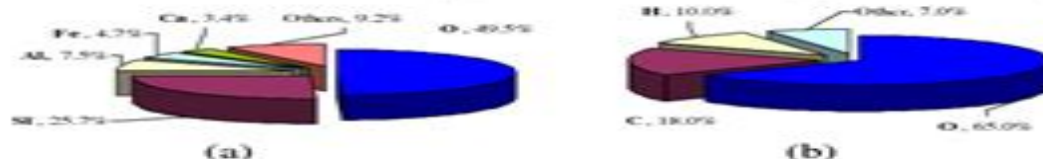
- form of matter that has distinct composition and properties

a. Elements

- Substance composed of one kind of atom and cannot be decomposed into simpler substances
- There are 118 known elements of which 91 elements exist in the Earth in macroscopic or trace quantities, 3 elements (Tc, Pm and Np) exist as a result of rare decay of heavy elements and 24 elements (Am to Uuo) are synthetically prepared

b. Compounds

- Substance formed when elements interact with other elements
- A pure compound is made up of the elements in the same proportion by mass – Law of Definite Proportions by Joseph Louis Proust (1754-1826)
- Abundance of the elements in (a) earth's crust and (b) human body



2. Mixture

- form of matter that consists of different substances and whose composition varies
- a. **Heterogeneous mixture** – mixture that does not have uniform composition, properties and appearance
- b. **Homogeneous mixture** – mixture that is uniform throughout

B. Properties of Matter

1. **Physical properties** – observed without changing the identity and composition of the substance
2. **Chemical properties** – observed when a substance react to form other substances
3. **Intensive properties** – properties independent of the amount of substance present such as melting point, density, temperature, etc.
4. **Extensive properties** – properties dependent of the amount of substance present such as mass and volume

C. Changes in Matter

1. **Physical change** – changes that does not result to formation of new substances
2. **Chemical change** – changes that result to formation of new substances

Atomic Theory

A. Dalton's Atomic Theory

- Proposed by John Dalton (1766-1844)

Inorganic Chemistry Study Guide

SB Merriam



Inorganic Chemistry Study Guide:

Whispering the Techniques of Language: An Emotional Quest through **Inorganic Chemistry Study Guide**

In a digitally-driven earth wherever displays reign supreme and immediate interaction drowns out the subtleties of language, the profound strategies and mental subtleties hidden within words often go unheard. However, set within the pages of **Inorganic Chemistry Study Guide** a interesting fictional prize pulsating with natural emotions, lies an exceptional journey waiting to be undertaken. Penned by a talented wordsmith, this wonderful opus attracts readers on an introspective trip, gently unraveling the veiled truths and profound affect resonating within the fabric of every word. Within the mental depths of the touching review, we shall embark upon a heartfelt exploration of the book is core themes, dissect their interesting publishing type, and succumb to the effective resonance it evokes strong within the recesses of readers hearts.

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