

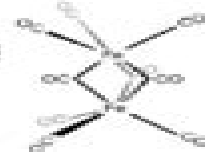
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CHEM 2218 Inorganic Chemistry I (Final Exam sample paper)

Part I: Multiple Choices (2 points each, 50 points in total)

- For a cube, the number of C_6 , C_4 , and C_2 rotational axes is
(a) 1, 3, and 4, respectively (b) 0, 2, and 4, respectively (c) 0, 3, and 4, respectively
- The CO diatomic molecule contains infinite number of
(a) σ_v (b) i (c) σ_h (d) C_2
- Identify the wrong label(s) in the following list used for representations of point groups.
 C_{1000} , C_{100} , D_{1000} , D_{2v} , C_4 , I , O_h , T_d
- The $\text{Fe}_2(\text{CO})_9$ molecule shown does not possess what element of symmetry?
(a) i (b) σ (c) S_6 (d) C_2
- A molecule having a point group of D_{6h} contains all the symmetry elements of a molecule having a point group of
(a) D_{3d} (b) D_{2d} (c) D_{3h} (d) none of the above
- How many mirror planes of symmetry are there in an icosahedron?
(a) 6 (b) 9 (c) 12 (d) 15 (e) 18
- Give the symmetry label for the molecular orbital associated with a trigonal planar structure (*you are given a character table for D_{3h}*).
(a) A_1^+ (b) A_1^-
(c) A_2^+ (d) A_2^-
- Identify the stronger acid?
 H_2Se vs. H_2S $[\text{Al}(\text{H}_2\text{O})_6]^{3+}$ vs. $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$
- Which one in the following is the least stable?
(a) $[\text{Al}(\text{H}_2\text{O})_6]^{3+}$ (b) $[\text{AlF}_6]^{3-}$ (c) $\text{Mo}(\text{H}_2\text{O})_6$ (d) $\text{Mo}(\text{CO})_6$
- Predict which way the following reactions will go (left or right)
(a) $[\text{AlF}_6]^{3-} + \text{Ga}(\text{CN})_3 \rightarrow [\text{GaF}_6]^{3-} + \text{Al}(\text{CN})_3$
(a) $\text{W}(\text{CO})_5(\text{PMe}_3) + \text{NMe}_3 \rightarrow \text{W}(\text{CO})_5(\text{NMe}_3) + \text{PMe}_3$
- Select the strongest base toward proton.
(a) NH_3 (b) NHMe_2 (c) PH_3 (d) PMe_3
- Which of the following reactions belong to the acid-base reactions?
(a) $\text{H} + \text{OH} \rightarrow \text{H}_2\text{O}$
(b) $\text{PF}_3 + \text{F}_2 \rightarrow \text{PF}_5$
(c) $\text{SiF}_4 + 2\text{NaF} \rightarrow \text{Na}_2[\text{SiF}_6]$
(d) $\text{Mo} + 6\text{CO} \rightarrow \text{Mo}(\text{CO})_6$



Inorganic Chemistry Question Papers

David Baud



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