

element	# valence e <sup>-</sup>	# e <sup>-</sup> moved	gained / lost	ion symbol	name
B	3 (group 13)	3	lost = 3 val e <sup>-</sup>	B <sup>+3</sup>	boron cation
F	7 (group 17)	3	gain = 3 val e <sup>-</sup>	F <sup>-1</sup>	fluoride anion
Li	1	1	lost	Li <sup>+1</sup>	lithium cation
N	5	3	gained	N <sup>-3</sup>	nitride anion
O	6	2	gained	O <sup>-2</sup>	oxide anion
Ca	2	2	lost	Ca <sup>+2</sup>	calcium cation
Br	7	1	gained	Br <sup>-1</sup>	bromide anion
S	6	2	gained	S <sup>-2</sup>	sulfide anion
Cl	7	1	gained	Cl <sup>-1</sup>	chloride anion
Ba	2	2	lost	Ba <sup>+2</sup>	barium cation
K	1	1	lost	K <sup>+1</sup>	potassium cation
Mg	2	2	lost	Mg <sup>+2</sup>	magnesium cation
Se	6	2	gained	Se <sup>-2</sup>	selenide anion
C (nonmet)	4	4	gained	C <sup>-4</sup>	carbide anion
Al	3	3	lost	Al <sup>+3</sup>	aluminum cation
Sn (met)	4	4	lost	Sn <sup>+4</sup>	tin (IV) cation or tin cation

# Monatomic Ions Packet Answer Key

**Paul B. Kelter**



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