

LONDON COLLEGE OF INTERNATIONAL CHEMISTS – 2020/2021

NAME: _____ COURSE: _____

ENTRY DESCRIPTION – A LEVEL 2020

ICM: 2020 (CONCEPTS IN ORGANIC CHEMISTRY)

CODE 11

Class : 2020-2021
 Date : 2020 - 2021

Page: No.

Page : 100 Words

PART – A

20 x 2 = 40 marks

Answer ALL the questions.

1. State three bonding principles.
2. Describe the structure of ethane in terms of orbital overlap.
3. Write the structure of ethane in terms of orbital overlap.
4. Describe the structure of ethane in terms of orbital overlap.
5. What is the structure of ethane in terms of orbital overlap?
6. Describe and explain the structure of ethane in terms of orbital overlap.
7. What are the three types of hybridization?
8. What are the three types of hybridization?
9. Describe the structure of ethane in terms of orbital overlap.
10. What are the three types of hybridization?

PART – B

20 x 2 = 40 marks

Answer any FIVE questions.

11. What are the three types of hybridization?
12. Describe the structure of ethane in terms of orbital overlap.
13. Describe the structure of ethane in terms of orbital overlap.
14. Describe the structure of ethane in terms of orbital overlap.
15. Describe the structure of ethane in terms of orbital overlap.
16. Describe the structure of ethane in terms of orbital overlap.
17. Describe the structure of ethane in terms of orbital overlap.
18. Describe the structure of ethane in terms of orbital overlap.
19. Describe the structure of ethane in terms of orbital overlap.
20. Describe the structure of ethane in terms of orbital overlap.
21. Describe the structure of ethane in terms of orbital overlap.
22. Describe the structure of ethane in terms of orbital overlap.

PART – C

20 x 2 = 40 marks

Answer any FIVE questions.

23. What are the three types of hybridization?
24. Describe the structure of ethane in terms of orbital overlap.
25. Describe the structure of ethane in terms of orbital overlap.
26. Describe the structure of ethane in terms of orbital overlap.
27. Describe the structure of ethane in terms of orbital overlap.
28. Describe the structure of ethane in terms of orbital overlap.
29. Describe the structure of ethane in terms of orbital overlap.
30. Describe the structure of ethane in terms of orbital overlap.
31. Describe the structure of ethane in terms of orbital overlap.
32. Describe the structure of ethane in terms of orbital overlap.

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Michael Seilmaier



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