Time: 3 hrs

Total Marks: 90

#### General Instructions:

- The question paper comprises of two sections, A and B. You are to attempt both the sections.
- There is no overall choice. However, internal choice has been provided in all the five questions of five marks category. Only one option in such question is to be attempted
- 3. All the guestions of Section-A and Section-B are to be attempted separately.
- Question numbers 1 to 3 in Section A are one mark questions. These are to be answered in one word or one sentence.
- Question numbers 4 to 7 in section A are two marks questions, to be answered in about 30 words each.
- Question number 8 to 19 in section-A are three marks questions, to be answered in about 50 words.
- Question number 20 to 24 in section-A are five marks questions, to be answered in about 70 words.
- Question numbers 25 to 42 in section-B are multiple choice questions based on practical skills. Each question is a one mark question. You are to select one most appropriate response out of the four provided to you.

#### SECTION-A

Attempt all questions from this section.

Q1. Why decomposition reactions are called the opposite of combination reactions?

Q2. What happens to the resistance of a conductor when its area of cross-section is increased?

Q3. What is tidal energy?

Q4. A solution of potassium chloride when mixed with silver nitrate solution, an insoluble white substance is formed. Write the chemical reaction involved and also mention the type of the chemical reaction?

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**Karin Nielsen-Saines** 

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