

10.1 Periodic Trends



10.1 PERIODIC TRENDS FROM CONCEPTS

Section 10.1 (continued)

In your notebook, read aloud the student periodicals table.

Copy the information in the table on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	21
_____	22
_____	23
_____	24

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It increases in the periodic table.
- _____ 20. It decreases in the periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are shiny and conduct electricity.
- _____ 23. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word as shown in smaller letters.

- _____ 24. There are more representative elements of elements.
- _____ 25. More than three-fourths of the elements in the periodic table are metals.
- _____ 26. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 27. Group 2A elements are the alkaline earth metals.
- _____ 28. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 29. Group 8A elements are inert nonreactive elements known as noble gases.
- _____ 30. Alkaloids have properties of both metals and nonmetals.

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