

Chemistry – Unit 4 Worksheet 4

DMC.5: *I can cite and interpret the evidence that supports the belief that some pure substances are made up of simpler particles (atoms) in a definite ratio.*

Answer the following questions. Be sure to show all mathematical work and reasoning and use complete sentences in explanations.

1. Two compounds of hydrogen and oxygen are tested. Compound I contains 120.0 g of oxygen and 15.0 g of hydrogen. Compound II contains 32.0 g of oxygen and 2.0 g of hydrogen.
 - a. Determine the ratio of the mass of oxygen to the mass of hydrogen in each of the compounds.
 - b. Why are the compounds not the same?
 - c. What is significant about these mass ratios?
 - d. If compound I is water, what could be the formula of compound II?
2. Nitrogen and oxygen combine to form a variety of compounds. The following data were collected for three different compounds of nitrogen and oxygen:

Analysis Data of Nitrogen & Oxygen Compounds	
Compound	Mass of Nitrogen that combines with every 1.0 g of Oxygen
A	1.750 g
B	0.8750 g
C	0.4375 g

- a. Additional evidence shows that the formula of compound B is NO. Sketch particle diagrams of molecules of all three compounds.
- b. Justify your representations above.
- c. There is no compound between Nitrogen and oxygen with a mass ratio of 2.00 grams of N for every 1.00 grams of oxygen. Why?

Modeling Chemistry Utest Answers

KJ Lindholm-Leary



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This book delves into Modeling Chemistry Utest Answers. Modeling Chemistry Utest Answers is a vital topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Modeling Chemistry Utest Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Modeling Chemistry Utest Answers
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- Chapter 3: Modeling Chemistry Utest Answers in Everyday Life
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- Chapter 5: Conclusion

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6. In chapter 5, this book will draw a conclusion about Modeling Chemistry Utest Answers. The final chapter will summarize the key points that have been discussed throughout the book.
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