Name \_\_\_\_\_

# Chemistry - Unit 7 Review

#### **Chemical Reaction Model**

 Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants

 $2 H_2 + O_2 \rightarrow 2 H_2O + energy$ 

Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

#### Applying the Model

Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the balanced equation for this reaction.

$$DB_2 + CD_3 + BAC \rightarrow C_2D_4 + AB_3$$

Which of these molecules would be called reactants?

# **Modeling Chemistry Unit 7**

**G** Orfield

**Modeling Chemistry Unit 7:** 

# Modeling Chemistry Unit 7 Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has be evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Modeling Chemistry Unit 7**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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