Name	
Date	Per

#### **Mixed Practice: Names and Formulas**

First determine whether each of the following compounds is *ionic* (M + NM), polyatomic (3 or more atoms), or covalent (2 NM's). Then write the name for each compound.

1) NO <sub>2</sub>		
2) NaBr		
3) SiO <sub>2</sub>		
4) P <sub>2</sub> Br <sub>4</sub>		
5) FeSO <sub>4</sub>		
6) SF <sub>6</sub>		
7) Li <sub>2</sub> S		
8) MgBr <sub>2</sub>		
9) N <sub>2</sub> S		
10) Be(OH) <sub>2</sub>		
11) SO <sub>3</sub>		
12) Cu <sub>2</sub> S		
13) BF <sub>3</sub>		
14) I <sub>2</sub>		
15) Pb <sub>3</sub> (PO <sub>3</sub> ) <sub>2</sub>		

# **Modeling Chemistry Unit 4 Answers**

**SA Dillow** 

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