

Unit 7 Worksheet 1: Molar-mass calculations

Four marks will be given for each question.

1. Write down balanced chemical equations

a) for the reaction of propanoic acid, acetic acid and acetic anhydride with calcium

b) for the reaction of calcium hydroxide with acetic anhydride.

2. Please call up hydrogen in 3

3. Hydrogen sulfide gas, which smells like rotten eggs, burns to give two products before it can be used as fuel. How many moles of oxygen are needed to completely burn 1 mole of hydrogen sulfide?

Equation: _____

Answer: _____

Change: _____

Notes: _____

4. Propane, C_3H_8 , burns in air to form carbon dioxide and water. If 10 moles of carbon dioxide are formed, how many moles of propane were burned?

Equation: _____

Answer: _____

Change: _____

Notes: _____

5. Acetic acid, CH_3COOH , has the ability to provide for creating hydrogens and the negative ions of hydroxide ions and peroxide. How many moles of acetic acid are needed to form 10 moles of hydroxide ions?

Equation: _____

Answer: _____

Change: _____

Notes: _____

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