

Chemistry – Unit 6 Notes 1 - Thomson Model of the Atom

J. J. Thomson performed experiments with cathode rays in an attempt to understand electricity – which was still a mystery in the late 1800s. Use the website *A Look Inside the Atom*¹ to find the conclusions that Thomson and other physicists drew regarding the mysterious cathode rays.

Thomson's 1897 Experiments - state the conclusions Thomson drew from each of his famous cathode ray experiments:

1. **First Experiment:** Thomson directed the beam at an electrometer and tried to separate the evidence of charge from the path of the beam. *What connection did Thomson find between charge and the cathode rays? Was the charge positive or negative?*
2. **Second Experiment:** Thomson tried passing the cathode ray through an electric field. *How did cathode ray beam behave when it passed through an electric field? What did he conclude after his second experiment?*
3. **Third Experiment:** Thomson did some careful measurements on how much the path of the cathode ray was bent in a magnetic field and how much energy they carried. From this work Thomson could describe the mass/charge ratio of the cathode ray particles. *What amazing result did Thomson find?*

Thomson's Atomic Model: Thomson presented three hypotheses from his experiments. Only two were accepted by physicists – in fact the third was shown to be wrong! From the first two came a model of the atom known as the *Plum Pudding* model. Complete the atom drawing below to illustrate Thomson's plum pudding model. Explain how this fits with his observations.



Modeling Chemistry U6 Ws1 V2 Answers

Marcel A. Müller



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Adopting the Song of Phrase: An Mental Symphony within **Modeling Chemistry U6 Ws1 V2 Answers**

In some sort of used by screens and the ceaseless chatter of instantaneous communication, the melodic elegance and emotional symphony produced by the published term usually fade into the background, eclipsed by the constant noise and distractions that permeate our lives. But, nestled within the pages of **Modeling Chemistry U6 Ws1 V2 Answers** a marvelous fictional value brimming with fresh thoughts, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, this fascinating masterpiece conducts viewers on a mental trip, skillfully unraveling the hidden songs and profound influence resonating within each cautiously constructed phrase. Within the depths of this touching examination, we can explore the book is main harmonies, analyze their enthralling writing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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