

## Chemistry – Unit 7 Review

### Chemical Reaction Model

1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

*In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants*



*Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.*

### Applying the Model

2. Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the **balanced equation** for this reaction.



Which of these molecules would be called reactants?

DB<sub>2</sub>, CD<sub>2</sub>, and BAC are reactants

# Modeling Chemistry U3 Review V2 2 Answers

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**Velavancorresponding**



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## **Modeling Chemistry U3 Review V2 2 Answers** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become much more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Modeling Chemistry U3 Review V2 2 Answers**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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