

Using Solubility Rules

- Soluble: will readily dissolve in water
- Insoluble: will not dissolve in water
- Slightly soluble: a tiny amount will dissolve in water

Soluble Compounds	Insoluble Compounds
NO_3^- , Na^+ , K^+ , NH_4^+	S^{2-} , CO_3^{2-} , PO_4^{3-}
Cl^- , Br^- , I^- (Ag^+ , Hg^{2+} , Pb^{2+})	OH^- (Na^+ , K^+ , Ca^{2+})
SO_4^{2-} (Ba^{2+} , Pb^{2+} , Ca^{2+})	

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The Princeton Review



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