

Stoichiometry Mixed Practice 1

1. How many grams of lead (II) nitrate are needed to produce 55.25 grams of sodium nitrate?



2. The average human requires 120.0 grams of glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) per day. How many grams of CO_2 are produced from this amount of glucose?



3. If there are 5.2×10^{25} atoms of calcium carbide (CaC_2), what mass of acetylene (C_2H_2) can be formed?



4. How many liters of oxygen at STP are needed to react with 5.2×10^{25} particles of hydrogen sulfide?



Stoichiometry Mixed Practice 2

1. How many grams of lead (II) nitrate are needed to produce 55.25 grams of sodium nitrate?



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Mixed Stoichiometry Practice

David Baud

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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