

## MOLES AND MASS

Name \_\_\_\_\_

Determine the number of moles in each of the quantities below.

1. 25 g of NaCl

\_\_\_\_\_

2. 125 g of  $\text{H}_2\text{SO}_4$

\_\_\_\_\_

3. 100. g of  $\text{KMnO}_4$

\_\_\_\_\_

4. 74 g of KCl

\_\_\_\_\_

5. 35 g of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

\_\_\_\_\_

Determine the number of grams in each of the quantities below.

1. 2.5 moles of NaCl

\_\_\_\_\_

2. 0.50 moles of  $\text{H}_2\text{SO}_4$

\_\_\_\_\_

3. 1.70 moles of  $\text{KMnO}_4$

\_\_\_\_\_

4. 0.25 moles of KCl

\_\_\_\_\_

5. 3.2 moles of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

\_\_\_\_\_

# Mixed Mole Problems Worksheet Answers Instructional Fair

**Frank Schaffer Publications, Joan  
DiStasio**



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