Practice: Balancing Equations #3

Show all work on a separate sheet of paper. Place coefficient final answers in the blanks provided. You can leave

². ____MgF₂+ ____Li₂CO₃
$$\Rightarrow$$
 ____MgCO₃+ ____LiF

3. ___
$$P_4$$
 + ___ O_2 \Rightarrow ___ P_2O_3

4. ____
$$RbNO_3$$
 + ____ BeF_2 \rightarrow ____ $Be(NO_3)_2$ + ____ RbF

5. ____ AgNO₃ + ____ Cu
$$\rightarrow$$
 ____ Cu(NO₃)₂ + ____ Ag

6. ____
$$CF_4 +$$
 ____ $Br_2 \Rightarrow$ ____ $CBr_4 +$ ____ F_2

7. ____ HCN + ___ CuSO₄
$$\Rightarrow$$
 ____ H₂SO₄ + ___ Cu(CN)₂

$$10.$$
___N₂+___NH₃

11. NaF + Br₂
$$\rightarrow$$
 NaBr + F₂

12.
$$Pb(OH)_2 + HCI \rightarrow H_2O + PbCl_2$$

13.____
$$AlBr_3 +$$
 ____ $K_2SO_4 \rightarrow$ ____ $KBr +$ ____ $Al_2(SO_4)_3$

14.___
$$CH_4+$$
 ___ $O_2 \rightarrow$ ___ CO_2+ ___ H_2O

15.____ Na₃PO₄+____ CaCl₂
$$\rightarrow$$
 ____ NaCl + ____ Ca₃(PO₄)₂

17.
$$AI + HCI \rightarrow H_2 + AICI_3$$

18.___
$$N_2 +$$
___ $F_2 \rightarrow$ ___ NF_3

19.___
$$SO_2 +$$
 ___ $Li_2Se \rightarrow$ ___ $SSe_2 +$ ___ Li_2O

20.
$$NH_3 + H_2SO_4 \Rightarrow (NH_4)_2SO_4$$

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RS Peters

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