

Chemistry Midterm Study Guide

Periodic Table of the Elements

Legend:

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Post-Transition Metals
- Metalloids
- Nonmetals
- Noble Gases
- Lanthanides
- Actinides

Electronegativity increases from left to right and bottom to top.

- Period** - is horizontally row of elements in the periodic table
ex - Chlorine like sodium and aluminum, shares the electron from
- Group** - is vertically row of elements in the periodic table
ex - Fluorine like potassium and calcium (transition) elements all elements in the same group
- Electronegativity** - ability to attract electrons (the more electronegative an element is, the more likely it is to attract electrons)
- Electropositivity** - The ability or tendency of an element to combine with oxygen and form an oxide, or more generally, to combine with other elements, such as, readily, electropositivity, tendency, reactivity

High School Chemistry Midterm Study Guide

Wenbin Ji

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