ncorrect):				
KCIO ₃ +	Heat →			
Cu +	FeBr ₃ (aq)	→		
C ₃ H ₆ +	O ₂ + Heat			14.5
K ₂ CO ₃	(aq) +	HNO₃ (aq) →		
(NH ₄) ₂ S (aq) +	NiCl2	(aq) →	
		not use "double re	placement)):	COLUMN TERMINA
) FeBr ₂ + 2 NH	r metathesis (do r l ₄ OH → Fe(OH) ₂	ot use "double re (s) + 2 NH ₄ Br	placement)):	
i) FeBr ₂ + 2 NH i) SO ₃ + H ₂ O - ii) Br ₂ + 2 Nal -	r metathesis (do r l ₄ OH → Fe(OH) ₂ + H ₂ SO ₄ - I ₂ + 2 NaBr	(s) + 2 NH ₄ Br	placement)):	
i) FeBr ₂ + 2 NH i) SO ₃ + H ₂ O - i) Br ₂ + 2 Nal - i) Ba(OH) ₂ + 2	metathesis (do r $_4$ OH \rightarrow Fe(OH) ₂ \rightarrow H ₂ SO ₄ \rightarrow I ₂ + 2 NaBr HNO ₃ \rightarrow Ba(NO:	(s) + 2 NH ₄ Br 3) ₂ + 2 H ₂ O	placement)):	
i) FeBr ₂ + 2 NH i) SO ₃ + H ₂ O - i) Br ₂ + 2 Nal - i) Ba(OH) ₂ + 2	metathesis (do r $_4$ OH \rightarrow Fe(OH) ₂ \rightarrow H ₂ SO ₄ \rightarrow I ₂ + 2 NaBr HNO ₃ \rightarrow Ba(NO:	(s) + 2 NH ₄ Br	placement)):	
i) FeBr ₂ + 2 NH ii) SO ₃ + H ₂ O - iii) Br ₂ + 2 Nal - ii) Ba(OH) ₂ + 2 iii) 2 H ₂ O ₂ + HE iii. 10 pts-	metathesis (do records) $_4$ OH \rightarrow Fe(OH) ₂ $_4$ H ₂ SO ₄ $_4$ I ₂ + 2 NaBr HNO ₃ \rightarrow Ba(NO ₃ AT \rightarrow 2 H ₂ O + C $_3$ A + 4B \rightarrow	3)2 + 2 H ₂ O 2(9) 2C + 3D + 3E	placement)):	
i) FeBr ₂ + 2 NH i) SO ₃ + H ₂ O - i) Br ₂ + 2 Nal - i) Ba(OH) ₂ + 2 ii) 2 H ₂ O ₂ + HE ii. 10 pts-	metathesis (do records) $_4$ OH \rightarrow Fe(OH) ₂ $_4$ H ₂ SO ₄ $_1$ 2 + 2 NaBr HNO ₃ \rightarrow Ba(NO ₃ AT \rightarrow 2 H ₂ O + C 3A + 4B \rightarrow ar masses are: A:	(s) + 2 NH ₄ Br (s) + 2 NH ₄ Br 3) ₂ + 2 H ₂ O	placement)):	

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