

Explain how you identified the limiting and excess reactant based only on your qualitative observations (not quantitative measurements or calculations). Type your answer into the box. Use complete sentences.

According to observation the initial color of bromocresol purple is yellow in color after adding the NaOH, it turns violet. After Changing the color yellow from purple indicates acid is completely consumed so excess of base addition changes the color into violet.

Table C. Calculations to Determine Theoretical Yields

<i>Description</i>	<i>Calculated Value</i>
Volume of NaOH solution	22.20 mL
Initial moles of NaOH	0.002451 moles
Initial moles of H_3PO_4	0.002602 moles
Mass of H_2O produced if NaOH runs out first	0.04414 g
Mass of H_2O produced if H_3PO_4 runs out first	1.265 g
Theoretical yield of H_2O (in grams)	1.265 g
Theoretical yield of Na_3PO_4 (in grams)	0.4267 g

Table D. Identification of Limiting and Excess Reactants from Calculations

<i>Description</i>	<i>Reactant Name or Formula</i>
Limiting Reactant	Phosphoric Acid
Excess Reactant	Sodium Hydroxide

Explain how you identified the limiting and excess reactant based on your calculations. Be specific. Type your answer into the box. Use complete sentences.

According to our calculation the reacting equivalent of H_3PO_4 is less than NaOH. Therefore H_3PO_4 is limiting reactant so product formation depends on H_3PO_4 because phosphoric acid is completely consumed.

Formal Lab Report On Limiting Reactant

Dr. Priyanka Gupta Manglik



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Calculating Theoretical Yield Is Not Impossible!

Michael S. Leonard, 2025-02-25 Calculating theoretical and percent yield is a fundamental skill for the laboratory This book primarily targets Organic Chemistry Laboratory courses at the high school or college and university level as a supplemental resource to help students master this skill It begins with simple examples from everyday life demonstrates the importance of balancing the equation addresses the role of the mole in these computations discusses different types of liquids considers the role of significant figures and culminates with the planning of syntheses There are suggestions for further reading as well as practice problems and questions to ensure mastery Begins with examples from everyday life that enable students to understand the concepts of theoretical and percent yield before applying those concepts to the laboratory Addresses the necessity of balancing the reaction equation the centrality of the mole in these calculations and the role of significant figures in reporting the answer Explains how to approach the calculations when using neat liquids or solutions The culmination of this text is the use of the same thought processes to plan the amounts of reactants needed for syntheses of desired quantities of product All of the problems in the book include detailed solutions with accompanying text to explain the answers and ancillaries also include suggestions for further reading Mine Drainage and Surface Mine Reclamation ,1988

Multiscale Operational Organic Chemistry John W. Lehman, 2009 This comprehensive laboratory text provides a

thorough introduction to all of the significant operations used in the organic lab and includes a large selection of traditional scale and microscale experiments and minilabs Its unique problem solving approach encourages students to think in the laboratory by solving a scientific problem in the process of carrying out each experiment The Second Edition contains a new introductory section Chemistry and the Environment which includes a discussion of the principles of green chemistry Several green experiments have been added and some experiments from the previous editions have been revised to make them greener

ChemCom American Chemical Society, 1993

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