

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that our "paper" writer is a chemist who has to write out what products may be made. What are the possibilities and how do you find them? The next 4 problems cover typical questions.

1. Chlorine reacts with bromine or iodine to produce bromine gas, Br_2Cl_2 and with some substances to produce perchlorine dioxide, Cl_2O_2 and a chlorine compound gas.

Reactants	10 g Cl_2	10 g Br_2	10 g I_2
Products	10 g	10 g	10 g
Excess	0 g	0 g	0 g

Chlorine gas
bromine gas

2. Silver reacts with displacement reactions, displacement metal reactions with hydrochloric acid to produce silver chloride and hydrogen gas. This is exactly what happens with hydrochloric acid. How much silver chloride is produced by reacting 10 g of silver with 10 g of hydrochloric acid?

Reactants	10 g Ag	10 g HCl
Products	10 g	10 g
Excess	0 g	0 g

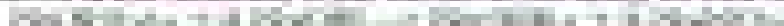
Silver chloride
hydrogen gas

3. Silver, Ag , reacts with hydrochloric acid to produce silver chloride and hydrogen gas. How much silver chloride is produced by reacting 10 g of silver with 10 g of hydrochloric acid?

Reactants	10 g Ag	10 g HCl
Products	10 g	10 g
Excess	0 g	0 g

How much silver chloride is produced by reacting 10 g of silver with 10 g of hydrochloric acid?

4. Chlorine reacts with silver to produce silver chloride and hydrogen gas. How much silver chloride is produced by reacting 10 g of silver with 10 g of hydrochloric acid?



How much silver chloride is produced by reacting 10 g of silver with 10 g of hydrochloric acid?

Reactants	10 g Ag	10 g HCl
Products	10 g	10 g
Excess	0 g	0 g

How much silver chloride is produced by reacting 10 g of silver with 10 g of hydrochloric acid?

Modeling Chemistry Unit 8 Answer

Raffaela Di Napoli



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