

## Chemistry – Unit 7 Worksheet 3

Write balanced chemical equations for the following reactions.

- a. Aluminum metal reacts with hydrochloric acid to form aluminum chloride.
- b. Acetylene ( $\text{C}_2\text{H}_2$ ), a hydrocarbon, burns with oxygen ( $\text{O}_2$ ) and carbon dioxide.
- c. Zinc reacts with copper(II) sulfate to form zinc sulfate and copper.
- d. Ethanol reacts with oxygen to form carbon dioxide and water.
- e. Sodium metal reacts with water to form sodium hydroxide.
- f. Nitrogen ( $\text{N}_2$ ) is placed in a hot glowing crucible and reacts with the gas.
- g. Magnesium metal burns when exposed to light to form magnesium hydroxide.
- h. Nitrogen gas reacts with some red glowing metal to form  $\text{Mg}_3\text{N}_2$ .

# Modeling Chemistry Unit 7 Ws3 Answers

**Lei Huang**

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