

Pre-Ap Chemistry Unit 6 – review for finals

These are the topics you should know and be able to answer questions about:

1. Chemical reactions

- What are some examples of evidence of chemical reactions?
 - Evidence that a chemical change *may have occurred* can include:
 - Production of bubbles (gas)
 - Formation of a precipitate (solid)
 - Production of heat and light
 - Color change
 - Noise
 - Ultimately, only the formation of a **NEW** substance can be absolute proof that a chemical change has occurred.
- What is the difference between chemical and physical reactions?
 - Chemical reactions produce new products – what you start with and end with are different
 - Physical reactions change the appearance of the substance, but you still have the same substance that you start with

2. Balance equations.

- Be able to balance equations
 - Either use math or tiddlywinks or draw a picture
- Why do we have to balance equations
 - Due to the law of conservation of mass. Mass is not created or destroyed during a chemical reaction. Due to this, the mass of the reactants must equal the mass of the products. Which means that there must be a balanced mass on each side of the reaction
- What are we trying to balance when we balance equations?
 - We need the same number of atoms of each element on each side of the arrow
- If you have balanced an equation and notice that the coefficients can be reduced, what do you? And why?
 - Reduce the coefficients.
 - We need the lowest ratio of coefficients
- Draw for me the particle representation of a balanced equation (circles) for the formation of water ($\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$)



- In order to balance the equation $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$ the correct coefficient for oxygen would be:
— i. 3

3. Types of reactions

- What are the different types of reactions?
 - Synthesis
 - Decomposition
 - Single replacement
 - Double replacement
 - Combustion

Modeling Chemistry Unit 6 Review V1

CL Gary



Modeling Chemistry Unit 6 Review V1:

Decoding **Modeling Chemistry Unit 6 Review V1**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Modeling Chemistry Unit 6 Review V1**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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