

- The element found in the periodic table in Group 6 and Period 5 is  
a. tungsten, W  
b. molybdenum, Mo  
c. tantalum, Ta  
d. tellurium, Te
- Which of the following groups is known for having both metallic and non-metallic properties?  
a. alkali metals  
b. alkali earth metals  
c. transition metals  
d. metalloids
- How many electrons does a strontium atom have in its outer shell?  
a. 1  
b. 2  
c. 7  
d. 8
- Which particles are responsible for the similar chemical and physical properties of elements in the same group?  
a. protons and neutrons  
b. number of energy shells  
c. valence electrons  
d. isotopes
- Which of the following chemical formulas most likely represents a pure substance containing ionic bonds?  
a.  $\text{Cl}_2$   
b.  $\text{SO}_2$   
c.  $\text{CO}_2$   
d.  $\text{BaO}$
- Consider the equation  $\text{X}_{\text{g}} + e^- \rightarrow \text{X}^- + \text{energy}$ . The "energy" term in the equation represents  
a. electronegativity  
b. electron affinity  
c. ionization energy  
d. nuclear charge
- Elements A, B, C, and D (found in Groups 1–17) have atomic radii of  $265 \mu\text{m}$ ,  $160 \mu\text{m}$ ,  $185 \mu\text{m}$ , and  $175 \mu\text{m}$ , respectively. Which element will most likely have the lowest ionization energy?  
a. A  
b. B  
c. C  
d. D
- List the following elements from smallest to largest atomic size:  
I. barium  
II. strontium  
III. bismuth  
IV. iodine  
a. III, I, II, IV  
b. IV, III, II, I  
c. IV, II, III, I  
d. IV, II, I, III
- Which shape and bond angle is predicted by VSEPR theory for  $\text{NF}_3$ ?  
a. trigonal pyramid, less than  $109.5^\circ$   
b. tetrahedral,  $109.5^\circ$   
c. trigonal planar,  $109.5^\circ$   
d. trigonal planar,  $120^\circ$

# Modeling Chemistry Unit 2 Test Answer Key

**Zhe Xu**



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