

1. Name these terms. Practice your knowledge by defining them in your own words.

a. mole

b. molar mass

c. Avogadro's number

d. empirical formula

e. molecular formula

2. Find the molar mass of the following:

a. KMnO_4 _____

d. oxygen gas _____

b. $\text{C}_2\text{H}_4\text{Cl}_2$ _____e. $\text{Ca}(\text{NO}_3)_2$ _____c. Ag_2CO_3 _____f. PbO_2 _____

3. Consider the masses of various isotopes below.

Type	Mass (g)	Relative mass
Hydrogen	1.74	1.00
Helium	3.18	
Carbon		12.00
Oxygen	16.04	

- a. Do the calculations necessary to complete the table.

- b. Explain the connections between these calculations and the atomic masses in the Periodic Table.

Modeling Chemistry U8 Review

M Walker



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