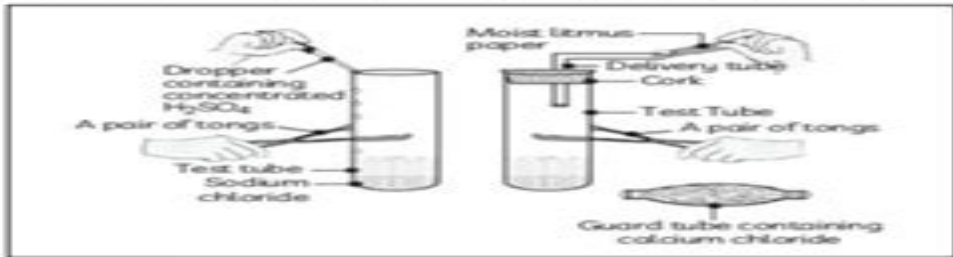
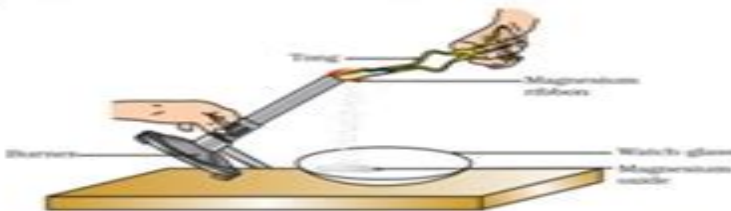


# SECTION - A

Select and write one most appropriate option out of the four options given for each of the questions 1 – 20

| Q. No  | Questions   | Marks                      |                        |                                  |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |
|--------|---|----------------------------|------------------------|----------------------------------|------------------------|----------------------------------|-----|------|--------|-------|------|-----|--------|--------|--------|------|-----|--------|-------|-------|--------|-----|------|-------|--------|--------|---|
| 1      | <p>The change in colour of the moist litmus paper in the given set up is due to</p> <div></div> <p>i. presence of acid<br/>ii. presence of base<br/>iii. presence of <math>H^+(aq)</math> in the solution<br/>iv. presence of Litmus which acts as an indicator<br/>(a) i and ii<br/>(b) Only ii<br/>(c) Only iii<br/>(d) Only iv.</p>  | 1                          |                        |                                  |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |
| 2      | <p>In the redox reaction<br/><math>MnO_2 + 4HCl \rightarrow MnCl_2 + 2H_2O + Cl_2</math><br/>(a) <math>MnO_2</math> is reduced to <math>MnCl_2</math> &amp; <math>HCl</math> is oxidized to <math>H_2O</math><br/>(b) <math>MnO_2</math> is reduced to <math>MnCl_2</math> &amp; <math>HCl</math> is oxidized to <math>Cl_2</math><br/>(c) <math>MnO_2</math> is oxidized to <math>MnCl_2</math> &amp; <math>HCl</math> is reduced to <math>Cl_2</math><br/>(d) <math>MnO_2</math> is oxidized to <math>MnCl_2</math> &amp; <math>HCl</math> is reduced to <math>H_2O</math></p>  | 1                          |                        |                                  |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |
| 3      | <div></div> <p>Which of the following is the correct observation of the reaction shown in the above set up?<br/>(a) Brown powder of Magnesium oxide is formed.<br/>(b) Colourless gas which turns lime water milky is evolved.<br/>(c) Magnesium ribbon burns with brilliant white light.<br/>(d) Reddish brown gas with a smell of burning Sulphur has evolved.</p>  | 1                          |                        |                                  |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |
| 4      | <p>With the reference to four gases <math>CO_2</math>, <math>CO</math>, <math>Cl_2</math> and <math>O_2</math>, which one of the options in the table is correct?</p> <table><tr><th>Option</th><th>Acidic oxide</th><th>Used in treatment of water</th><th>Product of respiration</th><th>Product of incomplete combustion</th></tr><tr><td>(a)</td><td><math>CO</math></td><td><math>Cl_2</math></td><td><math>O_2</math></td><td><math>CO</math></td></tr><tr><td>(b)</td><td><math>CO_2</math></td><td><math>Cl_2</math></td><td><math>CO_2</math></td><td><math>CO</math></td></tr><tr><td>(c)</td><td><math>CO_2</math></td><td><math>O_2</math></td><td><math>O_2</math></td><td><math>CO_2</math></td></tr><tr><td>(d)</td><td><math>CO</math></td><td><math>O_2</math></td><td><math>CO_2</math></td><td><math>CO_2</math></td></tr></table> | Option                     | Acidic oxide           | Used in treatment of water       | Product of respiration | Product of incomplete combustion | (a) | $CO$ | $Cl_2$ | $O_2$ | $CO$ | (b) | $CO_2$ | $Cl_2$ | $CO_2$ | $CO$ | (c) | $CO_2$ | $O_2$ | $O_2$ | $CO_2$ | (d) | $CO$ | $O_2$ | $CO_2$ | $CO_2$ | 1 |
| Option | Acidic oxide  | Used in treatment of water | Product of respiration | Product of incomplete combustion |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |
| (a)    | $CO$  | $Cl_2$                     | $O_2$                  | $CO$                             |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |
| (b)    | $CO_2$  | $Cl_2$                     | $CO_2$                 | $CO$                             |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |
| (c)    | $CO_2$  | $O_2$                      | $O_2$                  | $CO_2$                           |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |
| (d)    | $CO$  | $O_2$                      | $CO_2$                 | $CO_2$                           |                        |                                  |     |      |        |       |      |     |        |        |        |      |     |        |       |       |        |     |      |       |        |        |   |

# Mock Test Papers For Class 10 Cbse

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