

Mixed Stoichiometry Practice

Name _____

Date _____ Period _____

Write and/or balance the following equations (remember the diatomic elements and to cross-cross charges for ionic compounds!!!). Use the mole ratios from the balanced equations to solve the following stoichiometry problems. Use units and labels in all conversions, and round your answer to sig figs.

1. Potassium chlorate decomposes into potassium chloride and oxygen gas.

Balanced Equation: _____

2. How many moles of oxygen are produced when 3.0 moles of potassium chlorate decompose completely?

3. Butane (C_4H_{10}) undergoes combustion.

Balanced Equation: _____ $C_4H_{10}(l) +$ _____ $O_2(g) \rightarrow$ _____ $CO_2(g) +$ _____ $H_2O(l)$

4. How many grams of CO_2 are produced when 68 g of O_2 are reacted with an excess of butane?

5. Water decomposes into hydrogen gas and oxygen gas by electrolysis.

Balanced Equation: _____

6. How many grams of hydrogen will be produced when 6.0 moles of oxygen are produced?

7. How many grams of water are required to produce 9.00 grams of hydrogen?

Mixed Stoichiometry Answers

James E. Brady



Mixed Stoichiometry Answers:

Mixed Stoichiometry Answers Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such may be the essence of the book **Mixed Stoichiometry Answers**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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