

24.9 mol

2. What volume would the CO_2 in Problem 1 occupy at STP?

$$24.9 \text{ mol} \times 22.4 = 557.76 \text{ L}$$

560 L

3. A sample of NH_3 gas occupies 75.0 liters at STP. How many molecules is this?

$$\frac{75.0 \text{ L}}{22.4} = 3.3482 \text{ mol} \times (6.02 \times 10^{23}) = 20.156 \times 10^{23}$$

4. What is the mass of the sample of NH_3 in Problem 3?

5. How many atoms are there in 1.3×10^{22} molecules of NO_2 ?

Mixed Practice Mole Problems Worksheet Answer

Virginia Di-Anna Sutula



Mixed Practice Mole Problems Worksheet Answer:

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

Prentice Hall Physical Science

Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **The Effective Use of Analogies on Solving Mole Problems in High School Chemistry** Virginia Di-Anna Sutula, 1988

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