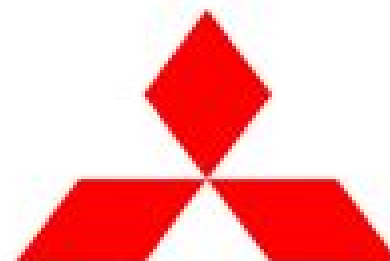




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 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$
 $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{H}_2\text{SO}_4 + 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$
br 1 1 n2 3 f2 2 nf3 br 2 2 c6h10 17 o2 12 co2 10 h2o br 3 1 hbr 1 khco3 1 h2o 1 kbr 1 co2 br 4 2 gabr3 3 na2so3 1 ga2 so3 3 6 nabr br 5 3 sno 2 nf3 3 snf2 1 n2o3 br using the following equation
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 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{H}_2\text{SO}_4 + 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$
gabr na2so3 gaz so3 3 nabr sno nf3 snf2

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answer $\text{CeSiO}_2 + 3\text{C} \rightarrow \text{SiC} + 2\text{CO}$

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 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{H}_2\text{SO}_4 + 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$
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