HYDROCARBONS



Hydrocarbon Compounds

Hydrocarbon compounds are compounds that consist only of carbon and hydrogen atoms. Hydrocarbons can be divided into 2, namely:

a. Saturated Hydrocarbons: single bonds

including saturated hydrocarbons:

- Alkane: open chain (C_nH_{2n+2})
- a Cycloalkane : closed chain (C_nH_{≥n})

b. Unsaturated Hydrocarbons: double bonds

- Alkene : double bond (C_nH_{2n})
- Alkyne : triple bond (C_nH_{2n-2})
- Alkadiene : two double bonds (C_nH_{2n-2})



Nomenclature of Alkanes

The nomenclature of alkanes is the basis for the nomenclature of other carbon compounds. Therefore the nomenclature of alkanes must be thoroughly understood.

Here are the names of askanes from C1-C10:

Formula	Name	Formula	Name
CH4	methane	C6H14	hexane
C ₂ H ₆	ethane	C7H16	heptane
C ₃ H ₈	propane	C ₈ H ₁₈	octane
C4H10	butane	C9H20	nonane
C ₅ H ₁₂	pentane	C ₁₀ H ₂₂	decane

Atkene names :

 C_2H_4 = ethylene

 C_3H_6 = propene

Hydrocarbons Note Taking Guide

AW Rasmussen

Hydrocarbons Note Taking Guide:

The book delves into Hydrocarbons Note Taking Guide. Hydrocarbons Note Taking Guide is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Hydrocarbons Note Taking Guide, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Hydrocarbons Note Taking Guide
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 - o Chapter 3: Hydrocarbons Note Taking Guide in Everyday Life
 - Chapter 4: Hydrocarbons Note Taking Guide in Specific Contexts
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- 6. In chapter 5, the author will draw a conclusion about Hydrocarbons Note Taking Guide. The final chapter will summarize the key points that have been discussed throughout the book.
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