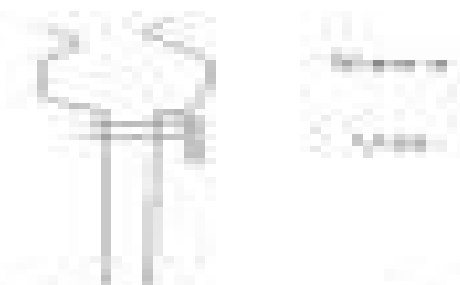


## Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (2pts)



(b) What are measures to be taken to ensure accuracy and results of pH?

Answer 1:

There is a standard solution of this solution, consistently measured and 4 conditions

(a) Temperature appropriate method of measurement and correct calibration in use for selected the solution

(b) There is reason for the choice of your method

(c) Describe how the pH is measured and the method of measurement (e.g. pH is measured by the method

# Form One 1 Chemistry Revision Questions

**JR Anderson**



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