

# EXPERIMENT: INVESTIGATING PRECIPITATION REACTIONS

PDF

## AIM

To compare the solubility of a range of ionic substances through precipitation.

## EQUIPMENT

☐ test tubes (6)

**Cations** (all 0.1 mol L<sup>-1</sup>) in dropper bottles:

- ☐ sodium nitrate, NaNO<sub>3</sub>
- ☐ silver nitrate, AgNO<sub>3</sub>
- ☐ lead(II) nitrate, Pb(NO<sub>3</sub>)<sub>2</sub>
- ☐ ammonium nitrate, NH<sub>4</sub>NO<sub>3</sub>
- ☐ magnesium sulfate, MgSO<sub>4</sub>
- ☐ iron(II) sulfate, FeSO<sub>4</sub>
- ☐ copper(II) sulfate, CuSO<sub>4</sub>
- ☐ zinc(II) sulfate, ZnSO<sub>4</sub>
- ☐ aluminium sulfate, Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>
- ☐ barium chloride, BaCl<sub>2</sub>
- ☐ calcium chloride, CaCl<sub>2</sub>
- ☐ iron(III) chloride, FeCl<sub>3</sub>

☐ test tube rack

**Anions** (all 0.1 mol L<sup>-1</sup>) in dropper bottles:

- ☐ potassium nitrate, KNO<sub>3</sub>
- ☐ potassium chloride, KCl
- ☐ sodium sulfate, Na<sub>2</sub>SO<sub>4</sub>
- ☐ sodium carbonate, Na<sub>2</sub>CO<sub>3</sub>
- ☐ sodium acetate, NaCH<sub>3</sub>COO
- ☐ sodium hydroxide, NaOH



**SAFETY** Wear safety glasses. Lead and barium salts are poisons so avoid contact with skin. If contact occurs wash thoroughly with soap and water. Dispose of as directed by your teacher. Consult "Chemical Safety in Schools" booklet (CSES).

## PROCEDURE

- 1 Put a 5 mL of a chemical containing each anion into each of six test tubes. Take care not to mix droppers.
- 2 Add 5 mL of a chemical containing the first cation to each. Record your observations.
- 3 Clean all test tubes thoroughly as incorrect results can arise from contamination of solutions.

# Experiment 17 Precipitation Reactions Answers

**R P Manchanda**



## **Experiment 17 Precipitation Reactions Answers:**

**Comprehensive Practical Science IX** S.P. Saxena,Dheeraj Saxena,2010      **AP Chemistry For Dummies** Peter J. Mikulecky,Michelle Rose Gilman,Kate Brutlag,2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam      Laboratory Manual for Science □ 9 A. K. Raj, Laboratory Manual for Science is a series of five books for classes 6 to 10 These are complimentary to the Science textbooks of the respective classes The manuals cover a wide range of age appropriate experiments that give hands on experience to the students The experiments help students verify scientific truths and principles and at the same time expose them to the basic tools and techniques used in scientific investigations Our manuals aim not only to help students better comprehend the scientific concepts taught in their textbooks but also to ignite a scientific quest in their young inquisitive minds      **Quantitative Analytical Chemistry** Hermenegild Arved Flaschka,Alfred James Barnard,Peter Earle Sturrock,1981      **NEET National Eligibility Cum Entrance Test Chemistry Class 12 Volume II** Priti Singhal,2024-11-12 The content in this book has been meticulously aligned with the latest NEET syllabus and trends reflecting insights from previous years papers We have focused on high yield topics and included chapter summaries topic wise MCQs and mock tests to provide a structured approach to learning Whether it s

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## **Table of Contents Experiment 17 Precipitation Reactions Answers**

1. Understanding the eBook Experiment 17 Precipitation Reactions Answers
  - The Rise of Digital Reading Experiment 17 Precipitation Reactions Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Experiment 17 Precipitation Reactions Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Experiment 17 Precipitation Reactions Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Experiment 17 Precipitation Reactions Answers
  - Personalized Recommendations
  - Experiment 17 Precipitation Reactions Answers User Reviews and Ratings
  - Experiment 17 Precipitation Reactions Answers and Bestseller Lists
5. Accessing Experiment 17 Precipitation Reactions Answers Free and Paid eBooks
  - Experiment 17 Precipitation Reactions Answers Public Domain eBooks
  - Experiment 17 Precipitation Reactions Answers eBook Subscription Services
  - Experiment 17 Precipitation Reactions Answers Budget-Friendly Options
6. Navigating Experiment 17 Precipitation Reactions Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Experiment 17 Precipitation Reactions Answers Compatibility with Devices
  - Experiment 17 Precipitation Reactions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Experiment 17 Precipitation Reactions Answers
  - Highlighting and Note-Taking Experiment 17 Precipitation Reactions Answers
  - Interactive Elements Experiment 17 Precipitation Reactions Answers
8. Staying Engaged with Experiment 17 Precipitation Reactions Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Experiment 17 Precipitation Reactions Answers
- 9. Balancing eBooks and Physical Books Experiment 17 Precipitation Reactions Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Experiment 17 Precipitation Reactions Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Experiment 17 Precipitation Reactions Answers
  - Setting Reading Goals Experiment 17 Precipitation Reactions Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Experiment 17 Precipitation Reactions Answers
  - Fact-Checking eBook Content of Experiment 17 Precipitation Reactions Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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