

Name: \_\_\_\_\_

## Double Replacement Lab

Reminder – Goggles must be worn at all times in the lab!

### PRE-LAB DISCUSSION:

A PRECIPITATE is an insoluble substance formed by the reaction between chemicals in solution. A list of reactions will be given to you. The reactants are on the left side of the equation, and the products are on the right. The precipitate will be found on the product side. Keep in mind that an equation gives a condensed picture of what happens when two chemicals react.

### PURPOSE:

The purpose of the lab is to observe semi-micro scale precipitation reactions, and to use a solubility table to identify the insoluble product, when there is one.

### PROCEDURE:

Predict the products and balance the following double replacement reactions. Then, identify the precipitates using the table on page 2. **You may want to do your work step on a scratch piece of paper and attach to lab later.**

| <u>Double Replacement Reactions</u> |                                 |   |                                   |   | <b>Precipitates<br/>(if any)</b> |
|-------------------------------------|---------------------------------|---|-----------------------------------|---|----------------------------------|
| A.                                  | NaCl                            | + | AgNO <sub>3</sub>                 | → |                                  |
| B.                                  | KI                              | + | AgNO <sub>3</sub>                 | → |                                  |
| C.                                  | NaOH                            | + | AgNO <sub>3</sub>                 | → |                                  |
| D.                                  | Na <sub>3</sub> PO <sub>4</sub> | + | AgNO <sub>3</sub>                 | → |                                  |
| E.                                  | NaCl                            | + | Pb(NO <sub>3</sub> ) <sub>2</sub> | → |                                  |
| F.                                  | KI                              | + | Pb(NO <sub>3</sub> ) <sub>2</sub> | → |                                  |
| G.                                  | NaOH                            | + | Pb(NO <sub>3</sub> ) <sub>2</sub> | → |                                  |
| H.                                  | Na <sub>3</sub> PO <sub>4</sub> | + | Pb(NO <sub>3</sub> ) <sub>2</sub> | → |                                  |
| I.                                  | NaCl                            | + | CuSO <sub>4</sub>                 | → |                                  |
| J.                                  | KI                              | + | CuSO <sub>4</sub>                 | → |                                  |
| K.                                  | NaOH                            | + | CuSO <sub>4</sub>                 | → |                                  |
| L.                                  | Na <sub>3</sub> PO <sub>4</sub> | + | CuSO <sub>4</sub>                 | → |                                  |

# Lab 10 2 Double Replacement Reactions

**G Psacharopoulos**



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**CHEMISTRY EXPERIMENTS** James Signorelli, 2014-09-19 Gifted and talented students and any student interested in pursuing a science major in college needs a rigorous program to prepare them while they are still in high school This book utilizes a format where the application of several disciplines science math and language arts principles are mandated Each lab concludes with either an essay or a detailed analysis of what happened and why it happened This format is based on the expectations of joining a university program or becoming an industrial science professional The ideal student lab report would be written in a lab research notebook and then the essay or final analysis is done on a word processor to allow for repeat editing and corrections The research notebook has all graph pages a title section and a place for the students and their assistants to sign and witness that exercise The basic mechanics of the lab report title purpose procedure diagrams data table math and calculations observations and graphs are handwritten into the book The conclusion is done on a word processor MS Word which allows the instructor to guide the student in writing and editing a complete essay using the MLA format When the final copy is completed the essay is printed and inserted into the lab notebook for grading At the end of the term the student has all their labs in one place for future reference These lab notebooks can be obtained for as little as 3 00 per book This is money well spent In our district the Board of Education buys the books for each student The BOE sees these books as expendable but necessary materials for all science and engineering instruction

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