



VHF FM TRANSCEIVER

IC-H16

UHF FM TRANSCEIVER

IC-U16

OWNER'S MANUAL

Icom Ic U16 User Guide

Robert R. Redfield



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World Radio TV Handbook ,1994 **Ham Radio** ,1990 **Ham Radio Magazine** ,1990 *Security* ,1990

Amateur Radio ,1998 *73 Amateur Radio Today* ,1996 **CQ** ,1997 **Passport to World Band Radio 1994**

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web figure page index 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95
82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and
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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in
the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the
condition at which a phase change occurs from liquid to vapour or vapour to liquid

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the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates
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