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ELECTROCHEMISTRY

Study Guide

A. Matching

Match each term in Column B to the correct description in Column A.

Column A

Column B

1. a conductor in a circuit that carries electrons to or from a substance other than a metal

✓ anode

2. one part of a voltaic cell, in which either oxidation or reduction occurs

✓ electrode

3. a voltaic cell in which the electrolyte is a paste

✓ cathode

4. the electrode at which reduction occurs

✓ dry cell

5. a voltaic cell in which a fuel undergoes oxidation and from which electrical energy is obtained continuously

✓ battery

6. an electrochemical cell that is used to convert chemical energy into electrical energy

✓ electrolysis

7. a tube containing a conducting solution

✓ voltaic cell

8. a group of cells that are connected together

✓ half-cell

9. the process in which electrical energy is used to bring about a chemical change

✓ salt bridge

10. the electrode at which oxidation occurs

✓ fuel cell

B. Multiple Choice

Choose the best answer and write its letter on the line.

11. In the electrolysis of brine, the substance produced at the cathode is

- a. chlorine c. sodium
b. oxygen d. hydrogen

12. Which of the following describes a dry cell?

- a. It can be recharged many times.
b. The graphite rod does not undergo reduction, even though it is the cathode.
c. It contains concentrated sulfuric acid.
d. all of the above

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