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CONCLUSION

What did you see in the experiment? If everything worked out, you should have noticed that the water changed color from red to blue as the water was heated. The color of the water is a function of the degree of heating. The color of the water is a function of the degree of heating.

When did you see in the experiment? If everything worked out, you should have noticed that the water changed color from red to blue as the water was heated. The color of the water is a function of the degree of heating. The color of the water is a function of the degree of heating.

When did you see in the experiment? If everything worked out, you should have noticed that the water changed color from red to blue as the water was heated. The color of the water is a function of the degree of heating. The color of the water is a function of the degree of heating.

think about this



Now, at the end of this section of the book, we have three chemical reactions that show the change in color of the water as it is heated. The color of the water is a function of the degree of heating. The color of the water is a function of the degree of heating.

ON YOUR OWN

1. What is the following element, period, and group in the periodic table of the elements?
2. Explain the difference between an atom and an element.

THE PERIODIC TABLE OF THE ELEMENTS

The periodic table of the elements (Figure 8.1) is a chemical's most important tool. It is more than just a listing of the elements. You see, as a chemist learned more about the behavior and properties of elements, they discovered they could arrange them based on what they were made of and how they behaved. Figure 8.1 shows that, instead of the table that you see here, it is a table of the elements. The elements are arranged in a table that is a function of the degree of heating. The color of the water is a function of the degree of heating.

FIGURE 8.1
The Periodic Table of the Elements

General Science Module 8 Study Guide Answers

Clifford Lane



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