

## Empirical and Molecular Formula Worksheet ANSWER KEY

Write the empirical formula for the following compounds.

- 1)  $C_6H_6$     **CH**
- 6)  $C_8H_{14}$      **$C_4H_7$**
- 7)  $WO_2$      **$WO_2$**
- 8)  $C_2H_6O_2$      **$CH_3O$**
- 9)  $X_{10}Y_{12}$      **$X_5Y_6$**
- 6) A compound with an empirical formula of  $C_2OH_4$  and a molar mass of 88 grams per mole. What is the molecular formula of this compound?     **$C_8O_4H_{16}$**
- 7) A compound with an empirical formula of  $C_4H_4O$  and a molar mass of 136 grams per mole. What is the molecular formula of this compound?     **$C_8H_8O_2$**
- 8) A compound with an empirical formula of  $CFBrO$  and a molar mass of 253.8 grams per mole. What is the molecular formula of this compound?     **$C_2F_2Br_2O_2$**
- 9) A compound with an empirical formula of  $C_3H_3N$  and a molar mass of 46 grams per mole. What is the molecular formula of this compound?     **$C_6H_6N_2$**
- 10) A well-known reagent in analytical chemistry, dimethylglyoxime, has the empirical formula  $C_2H_5NO$ . If its molar mass is 116.1 g/mol, what is the molecular formula of the compound?     **$C_8H_{10}N_2O_2$**
12. A certain blue solid contains 36.84% N and 63.16% O. What is the empirical formula of this compound? **The ratios are  $N_{1.50}O_{1.50}$ . Since 1.50 is not close to a whole number, we multiply *both* subscripts by 2. The empirical formula is thus  $N_2O_2$ . (The name is dinitrogen trioxide.)**
13. A sample of indium chloride weighing 0.5000 g is found to contain 0.2404 g of chlorine. What is the empirical formula of the indium compound?     **$InCl_2$**
14. An unknown compound was found to have a percent composition as follows: 47.0 % potassium, 14.5 % carbon, and 38.5 % oxygen. What is its empirical formula? If the true molar mass of the compound is 166.22 g/mol, what is its molecular formula?     **$K_2C_3O_4$**
15. Rubbing alcohol was found to contain 60.0 % carbon, 13.4 % hydrogen, and the remaining mass was due to oxygen. What is the empirical formula of rubbing alcohol?     **$C_3H_8O$**

# Empirical And Molecular Answer Key

**Sir Tarun Rupani**



## **Empirical And Molecular Answer Key:**

## **Empirical And Molecular Answer Key** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Empirical And Molecular Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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### **Table of Contents Empirical And Molecular Answer Key**

1. Understanding the eBook Empirical And Molecular Answer Key
  - The Rise of Digital Reading Empirical And Molecular Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Empirical And Molecular Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Empirical And Molecular Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Empirical And Molecular Answer Key
  - Personalized Recommendations
  - Empirical And Molecular Answer Key User Reviews and Ratings

- Empirical And Molecular Answer Key and Bestseller Lists
- 5. Accessing Empirical And Molecular Answer Key Free and Paid eBooks
  - Empirical And Molecular Answer Key Public Domain eBooks
  - Empirical And Molecular Answer Key eBook Subscription Services
  - Empirical And Molecular Answer Key Budget-Friendly Options
- 6. Navigating Empirical And Molecular Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Empirical And Molecular Answer Key Compatibility with Devices
  - Empirical And Molecular Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Empirical And Molecular Answer Key
  - Highlighting and Note-Taking Empirical And Molecular Answer Key
  - Interactive Elements Empirical And Molecular Answer Key
- 8. Staying Engaged with Empirical And Molecular Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Empirical And Molecular Answer Key
- 9. Balancing eBooks and Physical Books Empirical And Molecular Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Empirical And Molecular Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Empirical And Molecular Answer Key
  - Setting Reading Goals Empirical And Molecular Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Empirical And Molecular Answer Key
  - Fact-Checking eBook Content of Empirical And Molecular Answer Key
  - Distinguishing Credible Sources

- 
13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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