

## Part I: Determining the calorimeter constant:

| Data                                                | Trial   | Trial   |
|-----------------------------------------------------|---------|---------|
| (a) Mass of empty Styrofoam cups                    | 8.6 g   | 8.6 g   |
| (b) Mass of cups + 70 mL water                      | 76.6 g  | 77.7 g  |
| (c) Mass of cups + 70 mL water + 30 mL hot          | 103.4 g | 104.0 g |
| (d) Initial temperature of water in the calorimeter | 18.5 °C | 21.0 °C |
| (e) Temperature of the boiling water bath           | 98.1 °C | 98.1 °C |
| (f) Final temperature of calorimeter + added        | 39.8 °C | 40.6 °C |

| Calculation                                                                                                  |          |     |
|--------------------------------------------------------------------------------------------------------------|----------|-----|
| (g) Mass of cool water in cups, $m_{CW} [b - a]$                                                             | 68.0 g   | g   |
| (h) Mass of added hot water, $m_{HW} [c - b]$                                                                | 26.8 g   | g   |
| (i) Temperature change of cool water in the calorimeter ( $\Delta T_{CW} = T_f - T_{i(CW)}$ )                | + 21.3 K | + K |
| (j) Temperature change of added hot water ( $\Delta T_{HW} = T_f - T_{i(HW)}$ )                              | - 58.3 K | - K |
| (k) Calculate the heat <i>lost</i> by the hot water<br>$\Delta H_{HW} = m_{HW} C_s \Delta T_{HW}$            | - 6530 J | - J |
| (l) Calculate the heat <i>gained</i> by the cold water<br>$\Delta H_{CW} = m_{CW} C_s \Delta T_{CW}$         | + 6050 J | + J |
| (m) Calculate the heat <i>gained</i> by the calorimeter<br>$\Delta H_{cal} = -\Delta H_{HW} - \Delta H_{CW}$ | + 480 J  | + J |
| (n) Calculate the calorimeter constant, $B$<br>$B = \Delta H_{cal} / \Delta T_{CW}$                          | 22.5 J/K | J/K |
| (o) Average calorimeter constant, $B$                                                                        |          | J/K |

# Lab 22 Specific Heat Of Metals Answers

**Peter Goodwin**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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